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Prentice Hall Chemistry Scientific Research Base Page 6 of 10 Assessment in Prentice Hall Chemistry The assessment strategies in Prentice Hall Chemistry will help both students and teachers alike ensure student success in content mastery as well as high-stakes test performance. A wealth of opportunities built into the Student

Section Summaries A two-page summary for each chapter in Prentice Hall Biology is also included in the first part of this Study Guide. The key concepts and vocabulary terms are summarized in an easy-to-read style. Use this portion of the Study Guide to review what you have read in every section of the textbook and to

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CHEMISTRY CHAPTER 5 NOTES 5.1 – Light and Quantized Energy • The Nuclear Atom and Unanswered Questions o Although Rutherford's scientific model of an atom was a breakthrough, it lacked detail about how electrons occupy the space surrounding the nucleus of an atom. o Questions Still Unanswered:

5.2 The Modern Periodic Table In the modern periodic table, elements are arranged by increasing atomic number (number of protons). Each row in the periodic table is called a period. The number of elements per period varies because the number of available orbitals increases from energy level to energy level. Each column on the periodic table is